This article was downloaded by: [Malmo Hogskola] On: 18 December 2011, At: 23:21 Publisher: Taylor & Francis Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Journal of Asian Natural Products Research

Publication details, including instructions for authors and subscription information: http://www.tandfonline.com/loi/ganp20

Chemical constituents of Equisetum debile

Jun-Ming Tan $^{\rm a}$, Yi-Hua Qiu $^{\rm a}$, Xin-Qi Tan $^{\rm a}$, Chang-Heng Tan $^{\rm b}$ & Kai Xiao $^{\rm c}$

^a PLA 98th Hospital, Huzhou, 313000, China

^b Department of Natural Medicinal Chemistry, Shanghai Institute of Materia Medica, Chinese Academy of Sciences, Shanghai, 201203, China

^c Laboratory of Toxicology & Pharmacology, Faculty of Naval Medicine, The Second Military Medical University, Shanghai, 200433, China

Available online: 10 Aug 2011

To cite this article: Jun-Ming Tan, Yi-Hua Qiu, Xin-Qi Tan, Chang-Heng Tan & Kai Xiao (2011): Chemical constituents of Equisetum debile, Journal of Asian Natural Products Research, 13:9, 811-816

To link to this article: <u>http://dx.doi.org/10.1080/10286020.2011.596829</u>

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: <u>http://www.tandfonline.com/page/terms-and-conditions</u>

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae, and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand, or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.



Chemical constituents of Equisetum debile

Jun-Ming Tan^a, Yi-Hua Qiu^a, Xin-Qi Tan^a, Chang-Heng Tan^{b*} and Kai Xiao^{c*}

^aPLA 98th Hospital, Huzhou 313000, China; ^bDepartment of Natural Medicinal Chemistry, Shanghai Institute of Materia Medica, Chinese Academy of Sciences, Shanghai 201203, China; ^cLaboratory of Toxicology & Pharmacology, Faculty of Naval Medicine, The Second Military Medical University, Shanghai 200433, China

(Received 2 April 2011; final version received 6 June 2011)

Three new compounds, debilitriol (1), debilignanoside (2), and equisetumine (3), along with nine known compounds, were isolated from the whole plant of *Equisetum debile*. Their structures were elucidated by spectral and chemical methods.

Keywords: Equisetum debile; debilitriol; debilignanoside; equisetumine

1. Introduction

Equisetum debile Roxb. (Equisetaceae) is a fern distributed widely in the south of China, Southeast Asia, and India, which has a long history as a Chinese folk medicine for the treatment of acute hepatitis, urethritis, conjunctivitis, and diarrhea [1]. Pharmacological studies disclosed that the alcohol extraction of E. debile could decrease the level of triglyceride and total cholesterol in rats and triglyceride in rabbits [2]. A few studies on *E. debile* [3-5] reported the existence of megastigmane glucosides, phenol glycosides, lignan glycosides, and flavonoids and their glycosides. As a continuous interest in Chinese folk medicinal plants, we investigated systematically the chemical constituents of the title plant, and obtained 12 compounds, including a new phenylhexane debilitriol (1), a new 8-O-4'neolignan glucoside debilignanoside (2), and a new alkaloid equisetumine (3) (Figure 1). The structures of these new compounds were determined to be (rel-2*R*,4*S*)-6-(3-hydroxy-4-methoxyphenyl)hexane-1,2,4-triol (1), (7*R*,8*S*)-guaiaglycerol- β -coniferyl ether 9-*O*- β -Dglucopyranoside (2), and 2-pentyl-1,5,9triazacyclotridecan-4-one (3), respectively, on the basis of spectral evidence.

2. Results and discussion

Debilitriol (1) was obtained as viscous syrup. Its negative and positive ESI-MS gave pseudomolecular ion peaks at m/z 255 $[M - H]^-$ and 279 $[M + Na]^+$, respectively, in agreement with the molecular formula $C_{13}H_{20}O_5$, which was also confirmed by the HR-ESI-MS at m/z 279.1205 $[M + H]^+$.

The ¹H, ¹³C NMR, and HSQC spectra of **1** showed ¹H and ¹³C signals for a 1,3,4trisubstituted aromatic ring at $\delta_{\rm H}$ 6.79 (1H, d, J = 8.2 Hz), 6.67 (1H, d, J = 2.0 Hz), and 6.63 (1H, dd, J = 8.2, 2.0 Hz), as well as $\delta_{\rm C}$ 147.9 (s), 147.7 (s), 137.3 (s), 121.0 (d), 117.1 (d), and 113.5 (d), a methoxyl at $\delta_{\rm H}$ 3.81 (3H, s) and $\delta_{\rm C}$ 57.1, four methylenes at $\delta_{\rm C}$ 68.4, 42.4, 41.8, and

*Corresponding authors. Email: chtan@mail.shcnc.ac.cn; kaixiaocn@gmail.com

ISSN 1028-6020 print/ISSN 1477-2213 online © 2011 Taylor & Francis DOI: 10.1080/10286020.2011.596829 http://www.informaworld.com

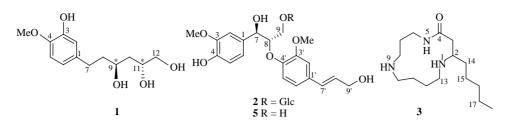


Figure 1. Structures of compounds 1-4.

32.9, and two oxygen-bearing methines at $\delta_{\rm C}$ 69.1 and 70.9, indicating a phenylhexane derivative. The aliphatic C6 moiety was elucidated to be -CH2CH2CH(OH)CH2 $CH(OH)CH_2OH$ based on the ${}^{1}H-{}^{1}H$ COSY cross-peaks of H₂-12/H-11, H-11/H₂-10, H₂-10/H-9, H-9/H₂-8, and H₂-8/H₂-7 (Figure 2). The MeO was assigned on C-4 of the aromatic ring by HMBC cross-peaks of MeO/C-4, H-2/C-4, and H-6/C-4 (Figure 2). The relative stereochemistry of 1 was determined to be 9,11-trans by the ¹H signals of 10-CH₂ and appeared as triplet (δ 1.52, t, $J = 6.2 \,\text{Hz}$) rather than two sets of multiplets [6]. The structure of 1 was, therefore, established to be (rel-2R,4S)-6-(3-hydroxy-4-methoxyphenyl)hexane-1,2,4-triol.

Debilignanoside (2) was obtained as a white amorphous powder. It had the molecular formula of $C_{26}H_{34}O_{12}$ as indicated by the HR-ESI-MS. The acid hydrolysis and GC analysis, together with NMR signals at $\delta_{\rm H}$ 4.27 (d, J = 7.7 Hz) and $\delta_{\rm C}$ 105.2 (d), 78.4 (2C, d), 75.7 (d), 72.0 (d), and 63.1(t), revealed a β -D-glucopyranosyl unit in the molecule of **2**.

In the ¹H NMR spectrum of 2, the aglycone moiety was observed to contain two 1,3,4-trisubstituted aromatic rings [δ 7.06 (d, J = 1.8 Hz), 6.88 (dd, J = 8.1, 1.8 Hz), and 6.74 (d, J = 8.1 Hz); 7.02 (d, J = 1.7 Hz), 6.98 (d, J = 8.3 Hz), and 6.90 (dd, J = 8.3, 1.7 Hz)], a pair of *E*-double bonds ($\delta 6.51$ and 6.24, $J_{\text{gemical}} = 15.7$ Hz), and two methoxyls (δ 3.85 and 3.81, each 3H, s). The ¹³C NMR spectrum of **2** showed 20 carbon signals $(2 \times \text{OCH}_3, 2 \times \text{OCH}_2,$ $2 \times OCH$, $8 \times = CH$, and $6 \times = C$) for the aglycone, which were almost superposed with those of guaiacylglycerol-β-coniferyl ether (4) [7] except for a downfield shift of C-9 ($\Delta \delta_{2-4} = 7.0$) due to the glycosidation effect. The above evidence indicated the structure of 2 to be a 9-O- β -D-glucopyranoside of 4, which was further substantiated by the HMBC experiment (Figure 2). The absolute configuration of 2 was determined to be 7R,8S on the basis of positive CD curve in the range of 210-250 nm [8] and the coupling constant of 5.4 Hz between H-7 and H-8 [9].

Equisetumine (3) was obtained as yellowish oil with positive Dragendorff's reaction. Its molecular formula was

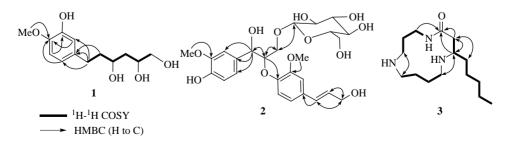


Figure 2. Selected ${}^{1}H-{}^{1}H$ COSY and HMBC correlations of 1-3.

determined to be C15H31N3O by the HR-EI-MS. The IR bands exhibited absorptions of NH (3408 and 3261 cm^{-1}) and CONH (1651 cm⁻¹). The ¹³C NMR spectrum displayed 15 carbon signals $(1 \times CH_3,$ $12 \times CH_{2}$, $1 \times CH$, and $1 \times \text{CON}$), indicating a monocyclic alkaloid. The HSQC and 1H-1H COSY spectra of 3 (Figure 2) revealed three isolated spin systems: (i) $-{}^{5}NH{}^{6}CH{}^{7}_{2}CH{}^{8}_{2}$ CH_2 , (ii) $-{}^{10}CH_2^{11}CH_2^{12}CH_2^{13}CH_2$, and (iii) $-{}^{3}CH_{2}^{2}CH(NH)({}^{14}CH_{2}^{15}CH_{2}^{16}CH_{2}^{17})$ $CH_2^{18}CH_3$)—. In the HMBC spectrum, marked correlations were observed at H₂-6/C-4, H₂-3/C-4, H-2/C-13, and H_2 -8/C-10 (Figure 2), indicating the three fragments to be connected through two N-atoms and an amide group to form a monocyclic structure.

Equisetumine is a rare macrocyclic polyamine alkaloid, biogenetically, which might be derived from spermidine (5) and γ -hydroxycaprylic acid (6) via a condensation reaction (Figure 3). Spermidine alkaloids were also reported from *Equise-tum palustre L.* [10], *Clerodendrum myricoides* [11], *Meehania fargesii* [12], *Dracocephalum tanguticum* [13], etc.

These known compounds were identified to be coumaric acid, *p*-hydroxybenzoic acid, ferulic acid, 5-hydroxymethyl-2-furfuraldehyde [14], equisetumoside B [15], guaiacylglycerol- β -coniferyl ether (4) [7], (+)-lariciresinol 9-*O*- β -D-glucopyranoside [16], (+)-isolariciresinol 3a-*O*- β -D-glucopyranoside [17], and thymidine [18] by comparison with published physical and spectroscopic data and/or with authentic samples.

3. Experimental

3.1 General experimental procedures

Optical rotations were determined on a PerkinElmer 341 polarimeter. The IR spectra were recorded on a Nicolet-Magna-750-FT-IR spectrometer. CD analysis was carried out on a Jasco J-810 spectropolarimeter. The NMR spectra were recorded on a Bruker AV-400 spectrometer with TMS as internal standard. ESI-MS and HR-ESI-MS were obtained on an Esquire 3000plus and a Q-TOF-Ultima mass spectrometers, respectively. EI-MS and HR-EI-MS spectra were obtained on a MAT-95 mass spectrometer. Silica gel (200-300 mesh, Qingdao Haiyang Chemical Co. Ltd, Qingdao, China), D-1400 macroporous resin (Yangzhou Pharmaceutical Factory, Yangzhou, China), RP-18 silica gel (150-200 mesh, Fuji Silysia Chemical Ltd, Aichi, Japan), and Sephadex LH-20 (Pharmacia Biotech AB, Uppsala, Sweden) were used for column chromatography (CC). Silica gel HSGF₂₅₄ (Yantai Jiangyou Guijiao Kaifa, Co., Yantai, China) was used for TLC. GC analysis was carried out on a PerkinElmer Sigma-115 gas chromatograph.

3.2 Plant material

The whole plants of *E. debile* were collected from Anji County, Zhejiang Province, China, in October 2009. The

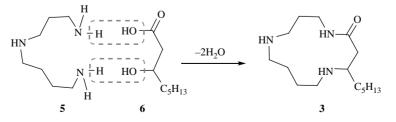


Figure 3. Possible biogenetic pathway of **3**.

plant was identified by Dr X.Q. Tan of PLA 98th Hospital. A voucher sample (No. 20091013) is deposited in the Herbarium of Shanghai Institute of Materia Medica, Chinese Academy of Sciences.

3.3 Extraction and isolation

The dried whole plant of E. debile (5 kg) was powdered and extracted with 25 liters 95% EtOH for three times. The concentrated extract was suspended in water and partitioned orderly with petroleum ether (PE), EtOAc, and BuOH. The EtOAcsoluble fraction (20 g) was subjected to CC of silica gel with gradient CHCl₃-MeOH (50:1, 25:1, 10:1, and 5:1) as eluents to give subfractions E1-E4. Fraction E1 offered 3 (40 mg) by repeated silica gel CC (PE-acetone, 5:1). E2 furnished 5hydroxymethyl-2-furfuraldehyde (90 mg) after purification of repeated silica gel CC (CHCl₃-MeOH, 20:1). Fraction E3 was separated into subfractions E3.1-E3.5 through a silica gel column with gradient CHCl₃–MeOH (15:1 \rightarrow 5:1) as eluents. E3.1 and E3.2 yielded solids, which were recrystallized to afford coumaric acid (18 mg) and *p*-hydroxybenzoic acid (13 mg), respectively. Compounds 4 (15 mg), 1 (5 mg), and ferulic acid (7 mg)were obtained from E3.3 after repeated CC of silica gel (CHCl₃-MeOH, 10:1).

BuOH fraction (37 g) was subjected to macroporous resin CC with gradient EtOH-H₂O (0, 20, 40, 60, and 95%) as eluents to gain B1-B5. Fraction B3 was separated into subfractions B3.1-3.5 by CC of RP-18 (60% MeOH). Fr.B3.2 yielded (+)-lariciresinol 9-O- β -D-glucopyranoside (25 mg) as solids. Fr.B3.3 offered 2 (18 mg) and equisetumoside B (11 mg), and Fr.B3.4 afforded (+)-isolariciresinol $3a-O-\beta-D-glucopyranoside$ (8 mg) after purification of Sephadex LH-20 CC (75% MeOH), respectively. Fraction B4 yielded thymidine (42 mg) as needles.

3.3.1 Debilitriol (1)

Colorless viscous syrup. $[\alpha]_{\rm D}^{24}$ -5.3(c = 0.417, MeOH). IR (KBr) $\nu_{\rm max}$: 3419, 1604, 1452, and 1512 cm^{-1} ; ¹H NMR spectral data (400 MHz, CD₃OD): δ 6.79 (d, J = 8.2 Hz, H-5), 6.67 (d, J = 2.0 Hz, H-2, 6.63 (dd, J = 8.2,2.0 Hz, H-6), 3.84 (m, H-11), 3.80 (m, H-9), 3.81 (s, 4-OMe), 3.47 (dd, J = 9.6, 5.0 Hz, H_a-12), 3.43 (dd, J = 9.6, 6.5 Hz, H_{b} -12), 2.66 (dt, J = 9.7, 7.7 Hz, H_{a} -7), 2.52 (dt, J = 9.7, 7.7 Hz, H_b-7), 1.70 (td, $J = 7.7, 6.3 \,\text{Hz}, \text{H}_2-8$, and 1.52 (t, $J = 6.2 \text{ Hz}, \text{ H}_2\text{--}10$). ¹³C NMR spectral data (100 MHz, CD₃OD): δ 147.9 (C-3, s), 147.7 (C-4, s), 137.3 (C-1, s), 121.0 (C-6, d), 117.1 (C-2, d), 113.5 (C-5, d), 70.9 (C-11, d), 69.1 (C-9, d), 68.4 (C-12, t), 57.1 (OMe, q), 42.4 (C-10, t), 41.8 (C-8, t), and 32.9 (C-7, t). ESI-MS: m/z 279 $[M + Na]^+$, 255 $[M - H]^-$. HR-ESI-MS: m/z 279.1205 $[M + Na]^+$ (calcd for C₁₃H₂₀O₅Na, 279.1208).

3.3.2 Debilignanoside (2)

White amorphous powder. $[\alpha]_{D}^{20}$ +5.5 (MeOH): (c = 0.583, MeOH). CD $\Delta \varepsilon_{215 \text{ nm}} + 2.5, \ \Delta \varepsilon_{225 \text{ nm}} + 4.7, \ \Delta \varepsilon_{240 \text{ nm}}$ $+ 3.5, \Delta \varepsilon_{248 \text{ nm}} + 4.7. \text{ IR (KBr) } \nu_{\text{max}}: 3419$ (OH), 1605, 1452, and 1512 (Ar) cm^{-1} . ¹H NMR spectral data (400 MHz, CD₃OD): δ 7.06 (d, J = 1.8 Hz, H-2), 7.02 (d, J = 1.7 Hz, H-2'), 6.98 (d, J = 8.3 Hz, H-2') 5'), 6.90 (dd, J = 8.3, 1.7 Hz, H-6'), 6.88 (dd, J = 8.1, 1.8 Hz, H-6), 6.74 (d,J = 8.1 Hz, H-5, 6.51 (br d, J = 15.7 Hz,H-7'), 6.24 (dt, J = 15.7, 5.6 Hz, H-8'), 4.97 (d, J = 5.4 Hz, H-7), 4.49 (td, J = 6.4)5.4 Hz, H-8), 4.27 (d, 7.7 Hz, H-1"), 4.19 $(2H, dd, J = 5.6, 1.1 Hz, H_2-9'), 3.85 (3H,$ s, MeO-3'), 3.83 (2H, d, J = 6.4, H₂-9), 3.81 (3H, s, MeO-3), 3.78 (dd, J = 11.6, $1.9 \text{ Hz}, \text{H}_{a}-6''), 3.64 \text{ (dd}, J = 11.6, 5.5 \text{ Hz},$ H_{b} -6"), and 3.40-3.15 (4H, m, H-3", H-4", H-2", and H-5"). 13 C NMR spectral data (CD₃OD, 100 MHz): δ 152.1 (C-3', s), 149.3 (C-4', s), 149.2 (C-3, s), 147.5 (C-4, s), 134.2 (C-1, s), 133.5 (C-1', s),

131.9 (C-7', d), 129.0 (C-8', d), 121.2 (C-6, d), 121.1 (C-6', d), 118.9 (C-5', d), 116.3 (C-5, d), 112.4 (C-2, d), 111.7 (C-2', d), 105.2 (C-1'', d), 85.2 (C-8, d), 78.4 (C-3'' and C-5'', each d), 75.7 (C-2'', d), 74.1 (C-7, d), 72.0 (C-4'', d), 69.5 (C-9, t), 64.2 (C-9', t), 63.1 (C-6'', t), 57.0 (MeO-3', q), and 56.9 (MeO-3, q). ESI-MS: m/z 561.1949 [M + Na]⁺ (calcd for C₂₆H₃₄ O₁₂Na, 561.1948).

3.3.3 Equisetumine (3)

Yellowish oil. $[\alpha]_{D}^{20} + 18$ (c = 0.7, CHCl₃). IR (KBr) *v*_{max}: 3408, 3261, 3076, 2929, 2658, 1651, 1556, 1441, 1367, 1178, 1066, and 629 cm^{-1} . ¹H NMR spectral data (CDCl₃, 400 MHz): δ 8.39 (1H, t, J = 5.7 Hz, NH-5), 3.73 (1H, m, H_a-6), 3.20–3.10 (4H, m, H-2, H_b-6, and H₂-8), 2.96 (1H, m, H_a-13), 2.92 (1H, m, H_a-10), 2.82 (1H, m, H_b-10), 2.62 (1H, m, H_b-13), 2.42 (1H, m, H_a-3), 2.38 (1H, m, H_b-3), 2.20–1.90 (3H, m, H₂-7 and H_a-11), 1.90–1.75 (4H, m, H_b-11, H₂-12, and H_a-14), 1.50-1.20 (7H, H_b-14, H₂-17, H₂-16, and H₂-15), and 0.87 (3H, t, J = 6.8 Hz, Me-18). ¹³C NMR spectral data (CDCl₃, 100 MHz): δ 173.0 (s, C-4), 55.3 (d, C-2), 49.3 (t, C-10), 48.9 (t, C-8), 44.2 (t, C-13), 41.6 (t, C-3), 38.4 (t, C-6), 31.94 (t, C-16), 31.90 (t, C-14), 26.7 (t, C-7), 26.1 (t, C-11), 26.0 (t, C-12), 24.7 (t, C-15), 22.6 (t, C-17), and 14.0 (q, C-18). EI-MS: *m/z* 269 (M⁺, 11), 252 (19), 226 (28), 198 (100), 155 (43), 140 (37), 126 (37), 100 (44), and 84 (70). HR-EI-MS: m/z 269.2455 $[M]^+$ (calcd for $C_{15}H_{31}N_3O$, 269.2467).

3.4 Acid hydrolysis of 2

Compound 2 (1 mg) was refluxed in 2 N HCl-dioxane (1:1 v/v, 2 ml) for 2 h. On cooling, the mixture was neutralized with NaHCO₃. After extraction with EtOAc, the aqueous layer was concentrated by blowing with N_2 . The residue was purified

by CC of Sephadex LH-20 (MeOH-H₂O 1:1, v/v) to give the sugar moiety. The purified sugar and standard D-glucose (Sigma-Aldrich, St Louis, MO, USA) were treated with L-cysteine methyl ester hydrochloride (2 mg) in pyridine (1 ml) at 60°C for 1 h. Then, the solution was treated with N,O-bis(trimethylsilyl)trifluoroacetamide (0.02 ml) at 60°C for 1 h. The supernatant was subjected to GC analysis to identify the sugars. Conditions for GC were capillary column, DB5-MS $(30 \text{ m} \times 0.25 \text{ mm} \times 0.25 \text{ }\mu\text{m})$, oven temperature program, 180-300°C at 6°C/min; injection temperature 350°C; carrier gas, He at 1 ml/min. D-Glucose was detected by comparing its retention time with that of the authentic sample ($t_{\rm R} = 12.30 \, {\rm min}$).

Acknowledgements

This study was financially supported by grants from the National Science & Technology Major Project 'Key New Drug Creation and Manufacturing Program' (Nos 2009ZX09301-001 and 2008ZXJ09002-011), and the National Science Natural Foundation of China (Nos 81072545 and 20872179).

References

- Compiler Group of State Administration of Traditional Chinese Medicine, *Material Medica of China* (Shanghai Science & Technology Press, Shanghai, 1999), Vol. 2, p. 395.
- [2] G.T. Wu, L. Xue, and Z.Q. Huang, J. Qiqihar Med. Coll. 25, 121 (2004).
- [3] X.H. Xu, C.H. Tan, S.H. Jiang, and D.Y. Zhu, *Helv. Chim. Acta* 89, 1422 (2006).
- [4] X.H. Xu, B.Q. Ruan, S.H. Jiang, and D.Y. Zhu, *Chin. J. Nat. Med.* 3, 93 (2005).
- [5] T.H. Thai, N.Q. Hung, C. Van Minh, N.X. Cuong, P.H. Yen, L.M. Huong, and P. Van Kiem, *Nat. Prod. Commun.* 3, 1903 (2008).
- [6] N. Harada, A. Saito, H. Ono, J. Gawronski, K. Cawronska, T. Sugioka, H. Uda, and T. Kuriki, *J. Am. Chem. Soc.* 113, 3842 (1991).
- [7] S.M. Li, K. Lundquist, and A. Wallis, *Phytochemistry* 49, 2125 (1998).
- [8] M.D. Greca, A. Molinaro, P. Monaco, and L. Previtera, *Phytochemistry* 35, 777 (1994).

- [9] S. Besombes, D. Robert, J.P. Utille, F.R. Taravel, and K. Mazeau, J. Agric. Food Chem. 51, 34 (2003).
- [10] M. Natsume and M. Ogawa, *Chem. Pharm. Bull.* **32**, 3789 (1984).
- [11] S. Bashwira and C. Hootele, *Tetrahedron* 44, 4521 (1988).
- [12] T. Murata, T. Miyase, T. Warashina, and F. Yoshizaki, J. Nat. Prod. 72, 1049 (2009).
- [13] S.Q. Wang, D.M. Ren, F. Xiang, X.N. Wang, C.J. Zhu, H.Q. Yuan, L.M. Sun, B.B. Lv, X.J. Sun, and H.X. Lou, *J. Nat. Prod.* **72**, 1006 (2009).

- [14] L.N. Fu, L.Y. Chen, R.H. Liu, and D.F. Chen, J. Chin. Med. Mat. 33, 1871 (2010).
- [15] J. Chang, L.J. Xuan, and Y.M. Xu, J. Integrative Plant Biol. 43, 193 (2003).
- [16] G.M. Della, A. Molinaro, P. Monoco, and L. Previtera, *Heterocycles* 36, 2081 (1993).
- [17] H. Achenbach, M. Lowel, R. Waibel, M. Gupta, and P. Solis, *Planta Med.* 58, 270 (1992).
- [18] D.Q. Yu and J.S. Yang, *Handbook of Analytical Chemistry*, 2nd ed. (Chemical Industry Press, Beijing, 1999), Vol. 7, p. 907.